**COSMOGENIC RADIONUCLIDES AND LIGHT NOBLE GASES IN EET 96026.** P. Ma<sup>1</sup>, G.F. Herzog<sup>1</sup>, L. Schultz<sup>2</sup>, H.W. Weber<sup>2</sup>, K. Knie<sup>3</sup>, T. Faestermann<sup>3</sup> and G. Korschinek<sup>3</sup>, <sup>1</sup>Dept. Chem. & Chem. Bio., Rutgers Univ., Piscataway, NJ 08854-8087, <sup>2</sup>Max-Planck-Institut für Chemie, Postfach 3060, D-55020 Mainz, Germany, <sup>3</sup>Fakultät für Physik, Technische Universität München, 85748 Garching, Germany.

**Introduction:** EET 96026 was first classified as an R3 chondrite but recent work indicates affinities to CO, CV or CK chondrites [1]. Noble gas contents suggest a short cosmic ray exposure [2]. We discuss the cosmic-ray exposure (CRE) history of this meteorite.

**Experimental methods and results:** A sample of 60 mg was dissolved with Be, Al, and Mn carriers in concentrated HF-HNO<sub>3</sub>-HClO<sub>4</sub>. We separated Cl from another aliquot of 23 mg. <sup>10</sup>Be, <sup>26</sup>Al and <sup>36</sup>Cl were measured at PRIME Lab, Purdue University; <sup>53</sup>Mn was measured as discussed previously [3,4]. The activities of <sup>10</sup>Be, <sup>26</sup>Al, <sup>36</sup>Cl and <sup>53</sup>Mn are 1.8±0.2, 8.9±0.5, 2.0±0.1 (dpm/kg) and

dpm/[kg Fe], respectively. Two samples with masses of 147 and 103 mg were analyzed for noble gases [5]. The contents of <sup>3</sup>He, <sup>4</sup>He and <sup>21</sup>Ne were reported in [2]. Additional results for those samples are as follows: <sup>20</sup>Ne (0.23/0.24), <sup>22</sup>Ne(0.09/0.09), <sup>36</sup>Ar (1.56/1.67), <sup>38</sup>Ar (0.31/0.33) and <sup>40</sup>Ar (249/226) in 10<sup>-8</sup>cm<sup>3</sup> STP/g.

**Discussion:** The high ratios of <sup>3</sup>He/<sup>21</sup>Ne (7.4) and <sup>22</sup>Ne/<sup>21</sup>Ne (1.21) indicate light shielding for EET 96026. CRE ages based on 1) the noble gas contents, 2) the elemental composition of CO chondrites [6], and 3) the method of [7] are  $T_3=0.28$ ,  $T_{21}=0.28$  and  $T_{38}$ =0.52 Ma. Assuming that EET 96026 was not huge in space, we expect the production rate of <sup>10</sup>Be, P<sub>10</sub> (atom/kg-min) to be bounded on the low side by 14 (for a radius, R<10 cm), and on the high side by 25 (for 40<R<50 and a depth, d>25). Accordingly, we find for a one-stage CRE age (Ma) 0.16<T $_{10}$ <0.30, and choose as our best estimate 0.23±0.07 Ma. Internal consistency then requires that  $P_{26}/P_{10}=2.45\pm0.05$ . From the galactic cosmic ray (GCR) production rate profiles of [8] and assuming CO composition, we would infer that our sample came from near the surface of a body with a radius of 15-40 cm. The <sup>53</sup>Mn activity gives a CRE age of 0.25-0.29 Ma for such a shielding condition. The activity of <sup>36</sup>Cl is lower than the expected value of 3.3 dpm/kg for a CRE age of 0.23 Ma, implying a terrestrial age of 0.21±0.03 Ma.

If we assume that  $T_{21}$ =0.28 Ma is a more precise estimate of the CRE age, then the production rate of  $^{10}$ Be was 15 atom/kg-min. According to [8],  $P_{10}$ =15 would be expected at the surface of a body with R<10 cm. The values of  $P_{26}$  and  $P_{53}$  calculated from the measured activities are about 30% larger than the modeled (GCR) production rates. This difference could reflect SCR production. Barring a very long terrestrial age, however, the low activity of  $^{36}$ Cl and the  $^{22}$ Ne/ $^{21}$ Ne ratio do not seem to support this interpretation.

In sum, EET 96026's transit time was similar to that of Isna (CO) and those of numerous CM chondrites [9], but atypically short for a CO, CV, or CK chondrite. Nominally larger CRE ages from noble gases than from <sup>26</sup>Al and <sup>10</sup>Be may reflect uncertainties or, less likely, some pre-exposure.

**References:** [1] Tonui, E.K. (2002) Pers. Comm. [2] Patzer A. et al. (1999) *LPC XXX*, 1145. [3] Knie K. et al. (2000) *Nucl. Instrum. Meth. Phys. Res B 172*, 717-720. [4] Schnabel C. et al. (2001) *LPS XXXII 1353*. [5] Schultz L. et al. (1991) *GCA 55*, 59-66. [6] Wasson J. and Kallemeyn G. (1988) *Phil. Trans. R. Soc. Lond. A* 325, 535-544. [7] Eugster O. (1988) *GCA 52*, 1649-1662. [8] Leya I. et al. (2000) *MPS 35*, 259-286. [9] Caffee M. and Nishiizumi K. (1997) *MPS 32*, *Suppl*, A26.